

IN THE SPECIFICATION:

Please insert the following paragraph at the beginning of the specification.

This application is a 371 of international application PCT/JP2004/004182, which claims priority based on Japanese patent application Nos. 2003-101226 filed April 4, 2003, and 2003-186632 filed June 30, 2003, which are incorporated herein by reference.

Please replace the paragraph beginning on page 10, line 9, with the following rewritten paragraph:

In this invention, the solvent containing an ester structure is preferably used, and a solvent containing a lactone structure is more preferable. The most preferable solvent is γ -butyrolactone. The boiling point used in this invention is the boiling point under one atmospheric pressure, i.e., the pressure of ~~$1.013 \times 10^5 \text{ N/m}^2$~~ $1.013 \times 10^5 \text{ N/m}^2$. Although the measurement of boiling point can be done by a well-known technique and it is not especially limited, it can be measured by using, for example, the boiling point meter of Swietoslowski.

Please replace the paragraph beginning on page 26, line 22,

with the following rewritten paragraph:

(2) Supposing the weight of substrate as W1, the weight of [[glass]] substrate and the dielectric composition as W2, the density of the dielectric composition as D, and the volume as V, the dielectric composition, $D = (W2 - W1) / V$.

Please replace the paragraph beginning on page 37, line 11, with the following rewritten paragraph:

Except that the solvent was propylene glycol ~~monomethyl~~ monomethylether acetate, a paste composition D-3 was prepared in the same way as that of the paste composition C-2. The boiling point of propylene glycol monomethyl acetate is 146°C. Then, according to the method of Example 1, a high dielectric constant composition was prepared and the result of evaluation of its dielectric characteristics is shown in Table 4. The relative dielectric constant was 46, the dielectric loss tangent was 4.7%, and the capacitance per area was 2.7 nF/cm², and was inferior in the electrical property. The porosity was 35 volume%.

Please replace the paragraph beginning on page 49, line 9, with the following rewritten paragraph:

A barium titanate filler (BT-05 of SAKAI CHEMICAL INDUSTRY

Co., Ltd., mean particle diameter: 0.5 μ m) 6067 weight parts, a ~~barium~~ strontium titanate filler (HPS-2000 of TPL. Inc., mean particle diameter: 0.045 μ m) 1613 weight parts, γ -butyrolactone 1523 weight parts and a dispersant (a copolymer having an acid group with a phosphoric-ester skeleton: BYK-W9010 of BYK-Chemie Japan KK) 77 weight parts were mixed and dispersed under ice-cooling for 1 hour using a homogenizer, and a dispersion liquid X-7 was obtained.

Please replace the paragraph beginning on page 52, line 14, with the following rewritten paragraph:

The epoxy resin (~~"Phenolite"~~ EPPN-502H of NIPPON KAYAKU CO., LTD.) 400 weight parts, a phenol novolak resin (TD-2131 of DAINIPPON INK AND CHEMICALS, Inc.) 400 weight parts and γ -butyrolactone 1000 weight parts were mixed and the resin solution Y-1 was obtained.

Table 1 beginning on page 63 has been amended as follows:

Paste composition								Dielectric Characteristics (1 MHz)				Film Characteristic
Examp e	Inorga nic filler	Resin	Curing agent	Solven t	Additi ve agent	Con ten t of the inorga nic filler in the solid conten t (wt%)	con ten t of the solven t in the paste (wt%)	Thickn ess (μm)	Releat ive dielec tric consta nt	Capaci tance (nF/cm ²)	Dieelc tric loss tangen t (%)	Porosi ty (volum e %)
1	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine	94	10	10	82	7.3	2.8	9

2	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine	94	15	15	73	4.3	3.4	12
3	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine	94	20	10	65	5.8	3.0	14
4	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine	94	25	8	58	6.4	3.2	20

5	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine BYK- W903	94	10	8	102	11.3	3.6	6
6	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Y- butyro lacton e	triphe nylpho sphine BYK- W903	94	15	10	95	8.4	3.1	7
7	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	N- methy l-2- pyrrol idone	triphe nylpho sphine	94	15	10	58	5.3	4.6	26

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8	Barium Titana te SAKAI CHEMIC AL BT- 05	Epoxy resin NIPPON KAYAKU EPPN50 2H	Phenol Novola k resin DAINIP PON INK TD2131	Ethyle ne glycol acetat e <u>diacet</u> <u>ate</u>	triphe nylpho sphine	94	15	10	64	5.7	4.8	21
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Table 8 on page 72 has been amended as follows:

Paste composition									
	Dispersi on liquid	Inorganic filler					Resin solutio n	Inorganic filler/re sin ratio	Stabilit y of dispersi on liquid
		Inorgani c filler composit ion	Mean particle diameter (µm)	Inorgani c filler composit ion	Mean particle diameter (µm)	Max/Min (ratio)			
Example 43	X-2	Barium Titanate	0.5	Barium Titanate	0.060	8.3	Y-1	79/21	Stabilit y
Example 44	X-7	Barium Titanate	0.5	Strontiu m Titanate	0.045	11.1	Y-1	79/21	Stabilit y
Example 45	X-8	Barium Titanate	0.5	Titanium Oxide	0.2	2.5	Y-1	[[81/29]] <u>81/19</u>	Instabil ity slightly (cohesio n)
Example 46	X-9	Lead type filler	0.9	Barium Titanate	0.059	15.3	Y-1	86/14	Stabilit y

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Comparative example 4	-							Y-1	0/100	-
Comparative example 5	X-10	Barium Titanate	0.5	-	-			Y-1	79/21	Stability
Comparative example 6	X-11	Barium Titanate	7	0.5	14			Y-1	79/21	Instability (filler sedimentation)
Comparative example 7	X-12	Barium Titanate	40	2.1	19			Y-1	79/21	Instability (filler sedimentation)
Comparative example 8	X-13	Barium Titanate	20	2.1	9.5			Y-1	79/21	Instability (filler sedimentation)
Comparative example 9	-	Barium Titanate	0.059	0.045	1.3			Y-1	79/21	Instability (cohesion)